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Development and evaluation of febuxostat solid dispersion through screening method

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1 **Development and evaluation of febuxostat solid**
2 **dispersion through screening method**
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4 **dispersion through screening method**

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24

25 **Abbreviations:**

26 Febuxostat (Febux)

27 Solid dispersion (SD)

28 Distilled water (DW)

29 Biopharmaceutics Classification System (BCS)

30 Differential scanning calorimetry (DSC)

31 Powder X-ray diffraction (PXRD)

32 Attenuated total reflectance Fourier transform infrared spectroscopy (ATR-FT-IR)

33 **ABSTRACT**

34 Febuxostat (Febux) is a BCS II drug and has a very low solubility. In order to overcome this
35 shortcoming, the purpose of study is to increase the *in vitro* dissolution (%) and drug release
36 (%) of Febux by using a screening method. The Febux-SD formulation was prepared by
37 screening solubilizers, pH agents, and carriers using with a solvent evaporation method.

38 The novel Febux SD formulation was successfully developed. The dissolution (%) of Febux
39 of optimal formulation (SD3) was higher than that of Feburic[®] tab in pH 1.2, distilled water
40 (DW), and pH6.8 buffer by 6.3-, 2.5-, and 1.1-fold, respectively, at 60 min. The *in vitro* drug
41 release (%) and permeability ($\mu\text{g}/\text{cm}^2$) of SD3 formulation were improved compared to those
42 of Feburic[®] tab in the pH shifting method and PBS (7.4), respectively. The SD3 formulation
43 was well maintained the stability for 6 months, and that of physicochemical properties were
44 altered. In conclusion, the Febux solubilization study with meglumine was first attempted and
45 successfully performed. Through the improved dissolution (%) of Febux, high bioavailability
46 of SD3 formulation is expected in animal and human studies.

47

48 **Keywords:** Febuxostat; meglumine; dissolution (%); *in vitro* drug release; stability; *in vitro*
49 permeability.

50 1. Introduction

51 Since 40% of existing drugs have low solubility, many pharmaceutical companies are
52 investing significant time, effort, and funds to solubilize poorly soluble drugs to increase oral
53 bioavailability. Thus, the research is being conducted on the solubilization of poorly soluble
54 drugs using various substances and methods. Drug solubilization methods include solid
55 dispersion (Sohn and Choi, 2022; Sohn et al., 2021a; Sohn et al., 2021b; Zaki et al., 2023),
56 complexation (Aung et al., 2022; Kamel et al., 2017; Volkova et al., 2021), self-micro and
57 nano-emulsifying formulations (Al-Amodi et al., 2020; Habib et al., 2021; Rangaraj et al.,
58 2019), micelles (Choi et al., 2020), and co-crystals (Jagia et al., 2022).

59 The model drug, febuxostat (Febux), which is commercial product as Uloric[®], is a new
60 bipurine-selective xanthine oxidase inhibitor. It is approved for treatment of hyperuricemia in
61 gout patients (Davoodi et al., 2020; Jagia et al., 2022). Uloric[®] (40 and 80 mg) tablets consist
62 of various excipients such as mannitol, lactose monohydrate, microcrystalline cellulose,
63 hydroxypropyl cellulose, and sodium croscarmellose in RxList. Febux belongs to the BCS class
64 II drugs with a pH-dependent solubility profile (weakly acidic drug) that exhibits improved
65 solubility at a basic pH (Yin et al., 2018). The BCS class II drugs have low solubility and high
66 intestinal permeability. These drugs can improve solubility, making them more readily
67 bioavailable than other BCS classes (Tambe et al., 2022).

68 The most recent research trends of Febux were SD formulations (Amin et al., 2020; El
69 Shenawy et al., 2019; Kaur et al., 2020; Moinuddin et al., 2020; Patel et al., 2021), as well as
70 self-micro and nano-emulsifying formulations (Al-Amodi et al., 2020; Habib et al., 2021;
71 Rangaraj et al., 2019), nanostructure lipid carriers (Varia et al., 2022), and co-crystal (Jagia et
72 al., 2022). Among them, the SD manufacturing method was selected owing to the reason that
73 our group has established a drug solubilization method with an SD manufacturing method, and
74 has an approach and manufacturing technology accordingly. Moreover, the SD manufacturing

75 method can ease obtaining a final product through solidification (Sohn and Choi, 2022). Also,
76 the solvent evaporation method in SD techniques was applied since it can be prepared at a low
77 cost and is time-efficient (requires only a stirrer and drying oven).

78 The most recent solubilization of the Febux formulation is as follows. Febux cocrystals
79 were prepared using the following methods; Febux and co-formers were mixed with
80 acetonitrile (50 μ L). The samples were dried at 80 °C for 2 h in a drying oven (Jagia et al.,
81 2022). Febux-SD formulations were prepared using different polymers (Kolliphor P[®] [188 and
82 237] and Eudragit RLPO). Then, SD manufacturing methods using hot melt (fusion), solvent
83 evaporation, and spray drying techniques were compared (Patel et al., 2021). Febux-co-
84 processed excipient formulations were prepared with crospovidone and microcrystalline blend
85 (1:1) with solvent evaporation method (Kaur et al., 2020). The Febux-SD formulations were
86 prepared using poloxamer (Kolliphor P[®] [188 and 237]). The samples were pasted using a
87 mixture of 50% (v/v) ethyl alcohol (5 mL) to obtain a paste and then dried in an oven at 60 °C
88 for 30 min (El Shenawy et al., 2019). The Febux- β -cyclodextrin-nanosponge formulations were
89 also prepared using the following method: β -cyclodextrin dissolved in dimethylformamide and
90 diphenyl carbonate was added to the reaction mixture at 100 °C for 4 h and then the white
91 powder was dried at 40 °C in an oven overnight and was subsequently ground in a mortar.
92 Febux was dissolved in dichloromethane, and then β -cyclodextrin-nanosponges were added,
93 following which the solution was pulverized until the dichloromethane evaporated (Amin et
94 al., 2020). As described above, Febux formulations have been developed by various methods.

95 The purpose of this study was to design the Febux-SD formulation with stable and
96 enhanced dissolution (%) of Febux. It is believed that novelty can be secured if a weak base
97 formulation that has not been performed in previous Febux studies is developed. It was
98 hypothesized that the dissolution (%) of Febux would improve if a weakly-basic substance was
99 used as a weakly-acidic drug (Febux). The Febux strategies are as follows: The solubilization

100 potential was first determined by measuring the solubility of Febux in various polymer
101 solutions (1%, w/v) and pH buffers. Second, formulations were developed by adding selected
102 weakly basic substances and solubilizers. Third, an optimal formulation was developed using
103 a carrier selection (Sohn and Choi, 2022). The prepared formulations were subjected to a pre-
104 dissolution test, and an optimal formulation was selected according to the results. The final
105 formulation was tested with dissolution, *in vitro* drug release, and permeability tests. Lastly,
106 physicochemical properties and stability were analyzed.

107 **2. Materials and Methods**

108 **2.1 Materials**

109 Febuxostat (Febux), meglumine, MgO, PVP/VA S-630, and Neusilin[®] (US2 and UFL2)
110 were provided by Yuyu Pharma Inc. (Korea, Suwon-si). Sodium oleate (extra pure) was
111 purchased from Junsei Chemical Co. Ltd. (Tokyo, Japan). Sodium hydroxide was purchased
112 from Daejung Chemicals & Metals Co. LTD (Siheung, Korea). Kolliphor[®] (P188 and P407),
113 PEG6000, Kollicoat[®] (IR), Kollidon[®] (K12, K17, K30, and K90), Kolliphor[®] (HS 15),
114 Soluplus[®], and TPGS were obtained from BASF (Ludwigshafen, Germany). Fumed silica
115 Aerosil[®] (200 and 300) were provided by Evonik (Essen, Germany). Mannitol, lactose,
116 microcrystalline cellulose, granular dicalcium phosphate anhydrate (DCP-A), and granular
117 dicalcium phosphate dihydrate (DCP-D) were obtained from Whawon Pharm (Seoul, Korea).
118 Sodium carbonate anhydrous, sodium bicarbonate, potassium hydroxide, sodium phosphate
119 dibasic anhydrous, ethyl alcohol, and universal buffers were purchased from Samchun Pure
120 Chemical Co., Ltd. (Pyeongtaek, Korea).

121 **2.2. Solubility test**

122 The reason for performing the Febux solubility test is to select choose which polymer and
123 pH substance to use. Thus, the solubility of Febux was evaluated in 1% (w/v) polymeric

124 solutions, DW, and universal buffers (pH 1.0, 4.0, 7.0, and 10.0)(Sohn and Choi, 2021b).
125 However, the pH4 buffer and Soluplus[®] prepared for the above solubility measurements were
126 excluded due to UV-vis interference. For this reason, a universal buffer (pH 4.0) was prepared
127 by the pH 4.0 buffer preparation method of Korean Pharmacopeia (edition 12).

128 Febux (10 mg) was put into the various solution and buffer solutions (10 mL) under the
129 stirring at 400 rpm using a multi-channel stirrer (MS-33MH, JEIO TECH, Korea) for 24 h at
130 37 ± 1 °C (n = 3). The sampling times were at 1, 2, 4, and 24 h and then assayed.

131 **2.3. Preparation of the Febux formulations**

132 The B (base) formulations (1-24) were prepared with various basic substances (including
133 meglumine, MgO, sodium oleate, sodium hydroxide, sodium carbonate anhydrous, sodium
134 bicarbonate, potassium hydroxide, and sodium phosphate dibasic anhydrous) using the solvent
135 evaporation method (**Table S1**). Briefly, Febux (40 mg), basic substances (different ratios of
136 Febux: basic substances = 1:1, 1:2, and 1:3), and ethyl alcohol (5 mL) were stirred at 400 rpm
137 for 30 min using a multi-channel stirrer at room temperature. The remaining solvent was dried
138 overnight at 80°C using an oven.

139 The F (polymer-base) formulations (1-12) were prepared based on the B2 formulation
140 (meglumine), and various polymers (including P188[®], P407[®], PEG6000, Soluplus[®], IR[®],
141 PVP/VA S-630, K12[®], K17[®], K30[®], K90[®], TPGS, and HS 15[®]), were prepared using the same
142 method as described above B formulations (**Table 1**). It was selected as the most used polymer
143 in pharmaceuticals and a nonionic polymer with good solubility in water.

144 The SD formulations (1–9) were prepared based on the F2 formulation and various carriers
145 (Aerosil[®] 200, Aerosil[®] 300, mannitol, Flowlac[®] 100, MCC, DCP-A, DCP-D, US2, and UFL2)
146 using the same method as described above F formulations (**Table 2**). Carriers most commonly
147 used in pharmaceuticals, such as mannitol, Flowlac[®] 100, MCC, and porous silica, direct

148 excipients, were selected. The prepared B, F, and SD formulations were passed through a 20-
149 mesh sieve (0.841 mm).

150 Additionally, experiments were performed to identify the major factors in the solubilization
151 of SD3 formulations (Sohn and Choi, 2022). This is done by excluding the excipient of the
152 SD3 formulation by one as follows. The SD3 (Febux, meglumine, P407[®], and mannitol in a
153 1:2:1:1 ratio), SD3-1 (Febux, meglumine, P407[®], and mannitol in a 1:0:1:1 ratio), SD3-2
154 (Febux, meglumine, P407[®], and mannitol in a 1:2:0:1 ratio), and SD3-3 (Febux, meglumine,
155 P407[®], and mannitol in a 1:2:1:0 ratio) were prepared in the same manner as described above
156 (Table 3).

157 2.4. UV-vis spectrophotometry

158 For all formulations, drug content (%), pre-dissolution (%), and dissolution (%) of Febux
159 were measured using a UV-vis spectrophotometer (X-ma 1000; Human Co., Korea) at 315 nm
160 (Amin et al., 2020; Patel and Thakkar, 2023). Undissolved Febux was separated by
161 centrifugation ($10,000 \times g$, 10 min) using a CF-10 microcentrifuge. The supernatants were
162 evaluated. All samples were measured in triplicate, and the measured absorbance was
163 calculated by a calibration curve (coefficient of determination, $R^2 = 0.9996$). The calibration
164 curve was calculated at 0.39–50 $\mu\text{g/mL}$ concentrations, and the samples were analyzed using
165 appropriate dilutions.

166 2.5. Pre-dissolution test

167 The convenient pre-dissolution test method was performed in DW for the selection of an
168 optimal Febux formulation using a multichannel magnetic stirrer (Sohn and Choi, 2022). The
169 Febux formulations (equivalent to 4 mg of Febux) were added to the beaker in DW (90 mL)
170 and stirred at 400 rpm at 37 ± 1.0 °C. All samples were withdrawn (1 mL) at 5, 15, 30, 45,
171 and 60 min.

172 **2.6. Dissolution test**

173 All samples (optimal formulation [SD3], physical mixture [PM3], pure Febux, Feburic[®]
174 tab [commercial product; replaced Uloric tab[®]], equivalent to 40 mg of Febux) were performed
175 using a dissolution tester (Distek 6300; New Brunswick, NJ, USA) in dissolution media (pH
176 1.2, DW, and pH 6.8 buffer [900 mL]) at 37 ± 0.5 °C and 100 rpm, in accordance with the
177 United States Pharmacopoeia Apparatus II paddle method (n = 6). The sampling times were 5,
178 15, 30, 45, and 60 min.

179 **2.7. *In vitro* release test**

180 An *in vitro* drug release study was performed using the pH-shifting method. The reason for
181 conducting this study is to predict the drug dissolution behavior *in vivo*. All samples (optimal
182 formulation [SD3], physical mixture [PM3], pure Febux, Feburic[®] tab [commercial product;
183 replaced Uloric tab[®]]) were pulverized in powder form using a mortar and pestle. An amount
184 equivalent to 4 mg of Febux was taken and dispersed in 5 mL of a pH 1.2 release media for
185 each sample. The subsequent method is the same as the previous study method (Sohn and Choi,
186 2021a).

187 **2.8. *In vitro* permeability test**

188 The *in vitro* permeability test was performed with various membranes in Franz diffusion
189 cell system (Bandtech Co., Ltd., Korea). Febux (equivalent to 0.4 mg/mL) was added to the
190 donor chamber, and then the samples were withdrawn at 1, 2, 4, 8, and 24 h. The protocol and
191 membranes were identical to the previous ones (Sohn and Choi, 2023).

192 **2.9. Physicochemical properties of the Febux-SD formulation**

193 Pure Febux, mannitol, meglumine, P407[®], PM3, and SD3 were thermally analyzed using a
194 DSC 60A (Shimadzu, Japan). The samples (2–5 mg) were put into the aluminum pan and then

195 sealed. The samples were heated from 5 °C to 350 °C at a scanning rate of 10 °C/min under a
196 nitrogen purge at 40 mL/min.

197 The chemical interactions of pure Febux, mannitol, meglumine, P407[®], PM3, and SD3
198 were confirmed using a FT-IR spectrometer (Nicolet6700, Thermo Scientific, USA). The range
199 is from 4,000 to 500 cm⁻¹ at a resolution of 2 cm⁻¹.

200 The crystallinities of pure Febux, mannitol, meglumine, P407[®], PM3, and SD3 were
201 analyzed using a high-resolution XRD (X'pert Pro MRD, PANalytical, The Netherlands). The
202 samples were scanned in 0.02° steps from 5° to 70° (diffraction angle 2θ) at 40 kV with 150-
203 mA Cu-Kα radiation.

204 **2.10. Stability**

205 Stability tests of the Febux formulations were performed using a pre-dissolution test for 1
206 and 6 months. Stability samples are placed in glass vials with caps in powder form and stored
207 under laboratory environmental conditions (temperature 20–25 °C, relative humidity 50–60%).
208 The Febux formulations (equivalent to 4 mg Febux) were put into the beaker in DW (90 mL)
209 and stirred at 400 rpm at 37 ± 1.0 °C for 60 min.

210 **2.11. Statistical analysis**

211 Statistical analysis was performed using the Student's *t*-test on SigmaPlot (ver. 12.5). Data
212 are presented as mean ± standard deviation (sd). In all analyses, $p < 0.005$ (***), $p < 0.01$ (**),
213 and $p < 0.05$ (*) were considered statistically significant.

214 3. Results

215 3.1 Characterization of the Febux formulations

216 3.1.1 Solubility test

217 Febuxostat (Febux) is a BCS II drug with good permeability and poor solubility in water.
218 However, permeability can further be increased with improved solubility. First, the solubility
219 test was conducted to determine the most elemental direction of the Febux formulations in
220 various 1% (w/v) polymeric solutions and pH buffers (**Fig. S1**). The solubility of Febux was
221 $32.5 \pm 15.5 \mu\text{g/mL}$ (P188[®]), $55.2 \pm 3.7 \mu\text{g/mL}$ (P407[®]), $12.7 \pm 1.9 \mu\text{g/mL}$ (PEG6000), $66.1 \pm$
222 $2.8 \mu\text{g/mL}$ (IR[®]), $17.6 \pm 1.7 \mu\text{g/mL}$ (PVP/VA S-630), $15.8 \pm 1.5 \mu\text{g/mL}$ (K12[®]), 16.1 ± 1.0
223 $\mu\text{g/mL}$ (K17[®]), $6.1 \pm 0.4 \mu\text{g/mL}$ (K30[®]), $28.3 \pm 1.6 \mu\text{g/mL}$ (K90[®]), $235.8 \pm 9.9 \mu\text{g/mL}$ (TPGS),
224 and $174.9 \pm 8.3 \mu\text{g/mL}$ (HS 15[®]). The solubility test results showed that TPGS has the highest
225 Febux solubility. In the pH buffers, the solubility of Febux were $0.2 \pm 0.1 \mu\text{g/mL}$ (pH 1), $1.9 \pm$
226 $0.1 \mu\text{g/mL}$ (pH 4.0), $911.4 \pm 59.1 \mu\text{g/mL}$ (pH 7.0), $1004.8 \pm 7.1 \mu\text{g/mL}$ (pH 10.0), and $31.3 \pm$
227 $1.6 \mu\text{g/mL}$ (DW) for 24 h. In conclusion, a change in pH environment is a crucial factor for
228 improving the solubility of Febux. When comparing the solubility of Febux in TPGS solution
229 (1%, w/v) and other polymer solutions (1%, w/v), the *p*-value was less than 0.005 (*t*-tests) for
230 24h. When comparing the solubility of Febux in pH buffer (pH 10) and other buffers and DW,
231 the *p*-value was less than 0.005 (*t*-tests) for 24h. Therefore, the B formulations were prepared
232 using basic substances.

233 3.1.2 Pre-dissolution test

234 The solubility test showed that Febux is more sensitive to pH than the polymer.
235 Formulations were prepared using various basic substances. The B formulations (1–24) were
236 fabricated with various basic substances at different ratios (**Table S1**). The B formulations (B1-
237 3) showing burst release showed a pre-dissolution (%) of over 90% at 60 min (**Fig. 1**). The pre-

238 dissolution (%) of the B3 formulation was more than 95% at 60 min. In general, the pre-
239 dissolution (%) of the B (1-3) formulations increased as the amount of meglumine increased.
240 Therefore, meglumine was selected as the basic substance, and the Febux to meglumine ratio
241 was determined for B2 (1:2) and B3 (1:3) formulations by comparing the initial and final pre-
242 dissolution (%) of Febux. Although the pre-dissolution (%) of B2 ($92.5 \pm 1.4\%$) and B3 (95.1
243 $\pm 2.1\%$) formulations were similar at 60 min, the B2 formulation was selected due to its lower
244 total weight in the formulation. When comparing the pre-dissolution (%) of Febux in the B2
245 formulation and other formulations, the *p*-value was less than 0.005 (*t*-tests) at 5 min, except
246 for the B3 formulation. With solubility and pre-dissolution tests, meglumine, a basic substance,
247 was chosen as the solubilizing agent of Febux.

248 The F formulations (1-12) were fabricated with various polymers (such as P188[®], P407[®],
249 PEG6000, Soluplus[®], IR[®], PVP/VA S-630, K12[®], K17[®], K30[®], K90[®], TPGS, and HS 15[®])
250 based on the B2 formulation (meglumine) (**Table 1**). The pre-dissolution (%) of Febux in the
251 F formulations were $91.3 \pm 0.7\%$ (F1), $97.0 \pm 2.8\%$ (F2), $86.1 \pm 1.1\%$ (F3), $87.7 \pm 2.0\%$ (F4),
252 $92.6 \pm 0.7\%$ (F5), $86.6 \pm 0.1\%$ (F6), $85.5 \pm 1.3\%$ (F7), $86.8 \pm 0.4\%$ (F8), $90.1 \pm 1.4\%$ (F9),
253 $86.4 \pm 0.3\%$ (F10), $91.4 \pm 0.4\%$ (F11) and $86.1 \pm 1.4\%$ (F12) at 60 min. Although the pre-
254 dissolution (%) of Febux decreased due to the addition of most polymers, F formulations (F1,
255 F5, F9, and F11) showed minor differences from the B2 formulation, and only the F2
256 formulation increased (**Fig. 2a**). Compared to the B2 formulation, the F2 formulation (P407[®])
257 improved the initial pre-dissolution (%) by approximately 4% and the final dissolution (%) by
258 5%. When comparing the pre-dissolution (%) of Febux in the F2 formulation and other
259 formulations, the *p*-value was less than 0.005 (*t*-tests) at 60 min.

260 The SD formulations (1–9) were prepared based on the F2 formulation (meglumine-P407[®])
261 with various carriers such as Aerosil[®] 200, Aerosil[®] 300, mannitol, Flowlac[®] 100, MCC, DCP-
262 A, DCP-D, US2, and UFL2 (**Table 2**). The pre-dissolution (%) of SD formulations were 93.2

263 $\pm 1.4\%$ (SD1), $95.3 \pm 2.1\%$ (SD2), $99.6 \pm 0.3\%$ (SD3), $95.3 \pm 1.6\%$ (SD4), $93.9 \pm 1.6\%$ (SD5),
264 $91.4 \pm 1.7\%$ (SD6), $87.1 \pm 1.5\%$ (SD7), $88.2 \pm 1.4\%$ (SD8), and $95.1 \pm 1.3\%$ (SD9) at 60 min
265 (**Fig. 2b**). Compared with the F2 formulation, the SD3 formulation (mannitol added) improved
266 by approximately 2% in the initial dissolution (%) and final dissolution (%). Compared to the
267 F2 formulation, there was no significant improvement of the SD3 formulation numerically
268 since it is challenging to increase further due to the high value of final dissolution (%) of Febux.
269 When comparing the pre-dissolution (%) of Febux in SD3 formulation and other formulations,
270 the *p*-value was less than 0.005 (*t*-tests) at 60 min. The SD3 formulation with the highest
271 dissolution (%) of the Febux was selected as the final formulation.

272 **3.2 Dissolution study**

273 The dissolution of Febux in SD3 formulation was compared with that of PM3, pure Febux,
274 and Feburic[®] tab (commercial product), as shown in **Fig. 3**. The dissolution methods for Febux
275 were the paddle method (II-method) and the dissolution medium (consist of 0.05 M phosphate
276 buffer [pH 6.0]) in US FDA dissolution method. The dissolution of Febux was conducted in a
277 commonly used media (pH 1.2, DW, and pH 6.8 buffer). Although Febux has been identified
278 as a pH-dependent drug through a solubility test, the solubilization effect at low pH must be
279 confirmed.

280 The dissolution (%) of samples in pH 1.2 media was low for pure Febux and Feburic[®] tab
281 (within 5%), but the dissolution (%) of PM3 and SD3 was $8.5 \pm 1.4\%$ and $30.7 \pm 6.5\%$,
282 respectively, at 60 min (**Fig. 3a**). At 5 min, SD3 showed higher dissolution (%) of Febux by
283 57.8-, 13.2-, and 5.0-fold compared to pure Febux, Feburic tab[®], and PM3 in pH 1.2 media,
284 respectively. At 60 min, SD3 showed higher dissolution (%) of Febux by 12.5-, 6.3-, and 3.5-
285 fold compared to pure Febux, Feburic[®] tab, and PM3, respectively, in pH 1.2 media. Although

286 the dissolution (%) of Febux in pH 1.2 media seemed low, SD3 showed a definite improvement
287 in dissolution (%) of Febux compared to that of the Feburic[®] tab.

288 The dissolution (%) of samples in DW was low for pure Febux (within 5%) and Feburic[®]
289 tab ($38.9 \pm 2.9\%$), while the dissolution (%) of Febux in PM3 and SD3 was $89.9 \pm 4.5\%$ and
290 $100.4 \pm 0.3\%$, respectively, at 60 min (**Fig. 3b**). The dissolution (%) of Febux in SD3 was
291 improved compared to that in PM3, Feburic tab[®], and pure Febux. At 5 min, SD3 showed
292 higher dissolution (%) of Febux by 105.4-, 3.9-, and 1.2-fold than pure Febux, Feburic[®] tab,
293 and PM3, respectively, in DW. At 60 min, SD3 showed 29.7-, 2.5-, and 1.1-fold higher
294 dissolution (%) of Febux than that of pure Febux, Feburic[®] tab, and PM3 in DW, respectively.

295 The dissolution (%) of samples in pH 6.8 buffer was pure Febux ($77.9 \pm 10.0\%$), Feburic[®]
296 tab ($88.9 \pm 7.4\%$), PM3 ($80.7 \pm 1.3\%$), and SD3 ($98.8 \pm 1.2\%$) at 60 min (**Fig. 3c**). The
297 dissolution (%) of Febux in SD3 was improved compared to that in PM3, Feburic tab[®], and
298 pure Febux. At 5 min, SD3 showed higher dissolution (%) of Febux by 1.3-, 1.1-, and 1.2-fold
299 compared to pure Febux, Feburic[®] tab, and PM3 in pH 6.8 media, respectively. At 60 min, SD3
300 showed 1.2-, 1.1-, and 1.2-fold higher dissolution (%) of Febux than pure Febux, Feburic[®] tab,
301 and PM3 in pH 6.8 media, respectively. In pH 6.8 buffer, the dissolution (%) of Febux in PM3
302 and SD3 tended to be slightly lower than in DW. The analysis results of SD3 were significantly
303 different ($p < 0.005$, Student's *t*-test) from those of the pure Febux, Feburic[®] tab, and PM3
304 formulations in all media at 5 and 60 min.

305 These results showed that the dissolution (%) of Febux in the SD3 formulation was
306 significantly improved compared to that of the Feburic[®] tab.

307 **3.3 *In vitro* drug release study**

308 The *in vitro* drug release profiles of pure Febux, Feburic[®] tab, PM3, and SD3 formulations
309 were evaluated in pH-shifting media such as SGF (pH 1.2) for 2 h and SIF (pH 6.8) for 48 h in

310 **Fig. 4.** The *in vitro* Febux release of samples in SGF media showed low release (%) as below
311 10%. It was confirmed that the dissolution (%) of Febux in SD3 at pH 1.2 increased compared
312 to that in other formulations. The *in vitro* Febux releases of samples in SGF media were 83.4
313 $\pm 7.2\%$ (pure Febux), $79.6 \pm 4.0\%$ (Feburic[®] tab), $76.4 \pm 4.9\%$ (PM3), and $95.2 \pm 7.2\%$ (SD3)
314 after 24 h. The *in vitro* Febux releases of samples in SGF media were $64.8 \pm 3.0\%$ (pure Febux),
315 $57.9 \pm 5.5\%$ (Feburic[®] tab), $61.5 \pm 5.8\%$ (PM3) , and $89.5 \pm 4.6\%$ (SD3) after 48 h. Overall, it
316 showed a pattern of increasing drug release up to 24 h and then decreasing for 48 h. In most
317 samples, drug release decreased by 15-20%, but SD3 decreased by 6%, and the decrease was
318 also low. These results confirmed that SD3 has superior stability in *in vitro* drug release media
319 compared to other formulations. As a result of comparing the *in vitro* drug release of Febux in
320 SD3 formulation and Feburic[®] tab and PM3 formulations, the *p*-value was less than 0.005 (*t*-
321 tests) at 24 h, except for pure Febux (*p* = 0.08). Through this *in vitro* release, the SD3
322 formulation is expected to have a higher dissolution (%) of Feburic[®] tab even *in vivo*.

323 **3.4. *In vitro* permeability study**

324 The samples as Feburic[®] tab, PM3, and SD3 formulations were evaluated with four types
325 of membranes in Franz diffusion cell system. In PVDF (hydrophilic and hydrophobic)
326 membranes, *in vitro* permeabilities in samples were $38.2 \pm 2.4 \mu\text{g}/\text{cm}^2$ and $16.5 \pm 3.1 \mu\text{g}/\text{cm}^2$
327 (Feburic[®] tab), $47.6 \pm 1.3 \mu\text{g}/\text{cm}^2$ and $15.2 \pm 2.3 \mu\text{g}/\text{cm}^2$ (PM3), and $54.3 \pm 2.9 \mu\text{g}/\text{cm}^2$ and
328 $22.9 \pm 1.2 \mu\text{g}/\text{cm}^2$ (SD3) for 24 h in **Fig 5 (a, b)**. According to these results, the SD3
329 formulation had 1.42-/1.38- and 1.14-/1.50- fold higher Febux permeability in PVDF
330 (hydrophilic and hydrophobic, respectively) than Feburic[®] tab and PM3. In PCTE (hydrophilic
331 and hydrophobic) membranes, *in vitro* permeabilities in samples were $38.5 \pm 1.7 \mu\text{g}/\text{cm}^2$ and
332 $40.6 \pm 3.1 \mu\text{g}/\text{cm}^2$ (Feburic[®] tab), $42.1 \pm 3.5 \mu\text{g}/\text{cm}^2$ and $44.3 \pm 0.6 \mu\text{g}/\text{cm}^2$ (PM3), and $54.2 \pm$
333 $3.6 \mu\text{g}/\text{cm}^2$ and $56.0 \pm 3.8 \mu\text{g}/\text{cm}^2$ (SD3) for 24 h in **Fig 5 (c, d)**. According to these results, the

334 SD3 formulation had 1.40-/1.37- and 1.28-/1.26- fold higher Febux permeability in PCTE
335 (hydrophilic and hydrophobic, respectively) than Feburic[®] tab and PM3. Moreover, except for
336 PVDF-hydrophobic, each formulation showed similar results in various membranes. The SD3
337 formulation showed the highest permeability in various membrane filters, and the *p*-value was
338 lower than 0.005 compared to other formulations.

339 3.5 Physicochemical properties

340 Pure Febux, excipients, PM3, and SD3 were analyzed using DSC to confirm the thermal
341 change in SD3, which is the optimal formulation. The samples of melting peaks were
342 confirmed at 209.7 °C (pure Febux), 169.3 °C (mannitol), 131.6 °C (meglumine), and 57.7 °C
343 (P407[®]). PM3 had melting peaks at 55.7 °C, 119.6 °C, and 154-163 °C (broad pattern), and
344 SD3 had melting peaks at 54.3 °C, 118.6 °C, and 154-163 °C (broad pattern), shown in **Fig.**
345 **6a**. As shown in the above, the melting peaks of SD3 and PM3 were observed in the order of
346 P407[®], meglumine, and mannitol, whereas the melting peak of pure Febux was not observed.
347 The results showed that it is difficult to distinguish the thermal changes in SD3.

348 The interaction between Febux and excipients in SD3 were performed using FT-IR (**Fig.**
349 **6b**).

350 The FT-IR spectra of Febux showed at 2229.3 (C≡N stretching), at 1674.5 cm⁻¹ (C=O
351 stretching), at 1604.7 cm⁻¹ and 1511.5 cm⁻¹ (C=C stretching), and at 1422.9 cm⁻¹(C-H
352 stretching). The FT-IR spectra of meglumine showed at 1074.2 cm⁻¹ (C–O), 1239.6 cm⁻¹ (C–
353 N) and at 2868.7 cm⁻¹ and 2918.4 cm⁻¹ (aliphatic C–H). The broad peaks at 3237.9 cm⁻¹ and
354 3316.0 cm⁻¹ were attributed to NH and OH stretching modes, respectively. The FT-IR spectra
355 of P407[®] showed at 1466.8 cm⁻¹ (C–H bending). A difference between PM3 and SD3 was that
356 C=O stretching band shifted from 1676.1 cm⁻¹ (PM3) to 1628.3 cm⁻¹ (SD3). This result
357 indicates that Febux and the excipient have hydrogen bonds. Thus, the dissolution (%) of Febux

358 in SD3 may have improved because of the chemical interactions between Febux and
359 meglumine.

360 The PXRD patterns of pure Febux, excipients, PM3, and SD3 are shown in **Fig. 6c**. The
361 peaks of pure Febux were identified at 6.5, 6.6, 12.7, 16.2, 19.8, 21.7, 23.7, 24.4, 25.7, and
362 25.8. The peaks of mannitol were identified at 10.3, 14.5, 18.6, 20.3, 20.9, 21.0, 23.3, 28.1,
363 29.3, 31.6, 32.4, 33.4, 35.9, and 38.6. The peaks of meglumine were identified at 8.9, 9.6, 12.3,
364 17.1, 17.9, 19.4, 21.9, 24.0, and 26.9. The peaks of P407[®] were identified at 18.9 and 23.1. The
365 peaks of PM3 were identified at 8.9, 12.3, 17.2, 17.9, 21.8, and 23.3. The peaks of SD3 were
366 identified at 8.9, 12.2, 17.9, 18.7, 21.7, and 23.3. The XRD patterns of PM3 and SD3 were
367 similar and mostly coincided with meglumine peaks. The peak intensity of Febux was
368 considered weak and could not be observed for PM3 and SD3.

369 **3.6 Stability study**

370 The pre-dissolution (%) of Febux in F and SD formulations was similar to that on the initial
371 day (within 3%), except for SD1 and SD2 (over 5%). Most formulations maintained stability
372 for 1 month (**Fig. S1**).

373 Composition of Febux-SD formulations after excluding each variable to identify the main
374 factors in **Fig. 7**. Initially, the pre-dissolutions (%) of formulations were $99.6 \pm 0.3\%$ (SD3),
375 $24.3 \pm 2.2\%$ (SD3-1), $93.8 \pm 3.1\%$ (SD3-2), and $97.0 \pm 2.8\%$ (SD3-3) at 60 min. The pre-
376 dissolutions (%) of formulations after 1 month were $98.7\% \pm 1.2\%$ (SD3), $22.4 \pm 3.9\%$ (SD3-
377 1), $92.3 \pm 1.1\%$ (SD3-2), and $93.5 \pm 1.6\%$ (SD3-3) at 60 min. The pre-dissolutions (%) of
378 formulations after 3 months were $99.4\% \pm 2.2\%$ (SD3), $22.6 \pm 3.9\%$ (SD3-1), $92.1 \pm 0.9\%$
379 (SD3-2), and $88.1 \pm 7.4\%$ (SD3-3) at 60 min. The pre-dissolutions (%) of formulations after 6
380 months were $99.6\% \pm 3.1\%$ (SD3), $21.7 \pm 2.2\%$ (SD3-1), $91.6 \pm 0.5\%$ (SD3-2), and $87.1 \pm 3.4\%$

381 (SD3-3) at 60 min. In most formulations, stability was maintained but tended to decrease by
382 approximately 10% in SD3-3.

383 Also, the crystallinity of Febux in SD3 was stable for 6 months, as shown in **Fig. 6c**. In
384 previous studies, the Febux-SD formulations did not undergo stability tests (Amin et al., 2020;
385 El Shenawy et al., 2019; Jagia et al., 2022; Kaur et al., 2020; Moinuddin et al., 2020; Patel et
386 al., 2021).

387 Additionally, solubility tests of pure Febux, Feburic[®] tab, PM3, and SD3 were performed.
388 The solubilities of the samples were 14.7 ± 3.1 $\mu\text{g/mL}$ and 37.1 ± 1.6 $\mu\text{g/mL}$ (pure Febux),
389 34.8 ± 3.9 $\mu\text{g/mL}$ and 153.5 ± 6.4 $\mu\text{g/mL}$ (Feburic[®] tab), 250.0 ± 23.8 $\mu\text{g/mL}$ and 871.1 ± 14.3
390 $\mu\text{g/mL}$ (PM3), 298.9 ± 28.7 $\mu\text{g/mL}$, and 949.0 ± 13.7 $\mu\text{g/mL}$ (SD3) for 1 h and 24 h, respectively,
391 indicating that the solubilities of SD3 were 25.5-fold (pure Febux), 6.2-fold (Feburic[®] tab), and
392 1.1-fold (PM3) higher than that of the other samples after 24 h. It was confirmed that SD3 is
393 superior to the other formulations regarding pre-dissolution (%), dissolution (%), and solubility.

394 4. Discussion

395 Through the solubility test, it was confirmed that Febux is pH-dependent (**Fig. S1**). Based
396 on these results, the development of Febux formulations using a basic substance that has yet to
397 be studied was selected as a strategy. The B (base) formulations (1-24) were developed with
398 eight basic substances. Meglumine was selected through screening of several types of basic
399 substance (**Fig. 1**). Comparing the pre-dissolution (%) of Febux in B2 formulation and other
400 formulations, the *p*-value was less than 0.005 (*t*-tests) at 5 min, except for the B3 formulation.
401 The basic substance was determined, and several types of polymers were added to develop the
402 F formulation (**Table 1**). The pre-dissolution (%) results of F formulations showed similar or
403 lower pre-dissolution (%) results to that of B2 formulation, showing unexpected results (**Fig.**
404 **2a**). Among the F formulations, only the F2 formulation to which P407[®] was added increased

405 pre-dissolution (%) of Febux. In a recent study on the solubilization of felodipine,
406 solubilization was successful using a P407[®]. In Ex vivo permeation study comparing pure
407 felodipine and felodipine SD-loaded rapidly dissolving oral films, the felodipine SD-loaded
408 rapidly dissolving oral films showed about 5 times higher permeability in the porcine buccal
409 mucosa (Sana et al., 2023). Although the drugs were different, the solubilizing effect of P407[®],
410 a solubilizing agent, was found. Depending on the properties of the polymer, the viscosity of
411 the formulations may increase to decrease dispersibility. Accordingly, it is considered that the
412 pre-dissolution (%) of Febux has decreased (Sohn et al., 2020a; Sohn et al., 2021b). **In Fig. 2b**,
413 according to the pre-dissolution (%) results, the SD3 (mannitol) increased the pre-dissolution
414 (%) of Febux by 2% compared to the F2 formulation. It is expected that there will be changes
415 due to its higher physicochemical interactions than Aerosil[®] and Neusilin[®], which have good
416 dispersibility.

417 The dissolution of Febux in the SD3 formulation was compared with that of PM3, pure
418 Febux, and Feburic[®] tab (commercial product), as shown in **Fig. 3**. In pH 1.2 media, the SD3
419 formulation showed higher dissolution (%) of Febux by 12.5-, 6.3-, and 3.5-fold compared to
420 pure Febux, Feburic[®] tab, and PM3, respectively. Although the low solubility at low pH due
421 to the characteristics of Febux was not clearly overcome, the improvement of the dissolution
422 (%) of Febux in the SD3 formulation was confirmed. In DW, The SD3 formulation showed
423 29.7-, 2.5-, and 1.1-fold higher dissolution (%) of Febux than that of pure Febux, Feburic[®] tab,
424 and PM3 in DW, respectively. The SD3 formulation showed superior dissolution (%) of Febux
425 increase compared to that of the Feburic[®] tab. Because the pH-shifting role of meglumine in
426 DW is clear, it appears that the dissolution (%) of Febux is greatly improved. In pH 6.8 buffer,
427 SD3 formulation showed 1.2-, 1.1-, and 1.2-fold higher dissolution (%) of Febux than pure
428 Febux, Feburic[®] tab, and PM3, respectively. The dissolution (%) of Febux in PM3 and SD3
429 tended to be slightly lower than that in DW. This may be because meglumine could not increase

430 the pH to the same extent as DW in pH 6.8 buffer. The initial and final dissolutions (%) of
431 Febux in the SD3 formulation were similar, because the Febux in the SD3 formulation showed
432 rapid wetting and dispersion (Sohn et al., 2020b). In previous studies, the dissolution (%) of
433 Febux in fast-dissolving tablets (Febux-co-containing blend crospovidone and microcrystalline
434 [1:1]) was approximately 87% in the pH 6.8 buffer (900 mL) for 10 min and after 30 min
435 decreased to 79%. It is considered to be due to the recrystallization of Febux in pH 6.8 buffer
436 (Kaur et al., 2020). Also, the dissolution (%) of the Febux-SD formulation (Febux: Kolliphor
437 P[®] 188 =1:1) was approximately 90% in a pH 7.4 buffer (900 mL) containing 0.35% w/v
438 Tween 20[®] for 60 min. However, the Febux-SD formulation showed a low Febux dissolution
439 (%) of approximately 50% at 5min (El Shenawy et al., 2019). The release of Febux- β -
440 cyclodextrin-nanosponges tablets showed a sustained release pattern in pH 6.8 buffer (900 mL)
441 for 10 min (Amin et al., 2020). Furthermore, the dissolution (%) of Febux co-crystals
442 (Febux:isonicotiamide = 2:1) was approximately 14.6% (13 μ g/mL) in pH 6.8 buffer (500 mL)
443 for 60 min (Jagia et al., 2022). The dissolution tests in the above studies were performed in pH
444 6.8 buffer or higher and showed a significantly lower dissolution (%) of Febux than that of our
445 SD3 formulation. These results showed that the dissolution (%) of Febux in the SD3
446 formulation was significantly improved compared to that of the Feburic[®] tab.

447 In **Fig. 4**, the *in vitro* Febux release of samples in SGF media showed low release (%) as
448 below 10%. It was confirmed that the dissolution (%) of Febux in SD3 at pH 1.2 increased
449 compared to that in other formulations. However, the low release (%) of Febux in SD3 was
450 considered to be due to the weak shear force compared with that of the dissolution test. Overall,
451 it showed a pattern of increasing drug release up to 24 h and then decreasing for 48 h. This is
452 likely due to due to the recrystallization of the drug in the release media. Through this *in vitro*
453 release, the SD3 formulation is expected to have a higher dissolution (%) of the Feburic[®] tab
454 even in *in vivo*.

455 In **Fig. 5**, the SD3 formulation had 1.40-/1.37- and 1.28-/1.26- fold higher Febux
456 permeability in PCTE (hydrophilic and hydrophobic, respectively) than Feburic[®] tab and PM3.
457 Although it is not an artificial barrier, SD3 showed superior results in four types of membranes.
458 Through this, intestinal permeability is also expected to be improved.

459 However, the limitation of this study is that it is challenging to prove changes in the Febux
460 of SD3 formulation in the body because there is no animal experiment. Therefore, in this paper,
461 it can be confirmed through previous research papers. In the first previous study, Febux-self-
462 nano-emulsifying formulations (Capmul MCM: Labrasol: Transcutol HP = 15:56.92:28.07 (%
463 w/w) as liquid, solid, and pellet formulations showed fast dissolution patterns similar to the
464 SD3 formulation. Also, oral bioavailability improved by 1.4- to 2.1-fold for the above three
465 formulations compared to pure Febux suspension in Sprague Dawley rats (Rangaraj et al.,
466 2019). In the second previous study, Febux-loaded β -cyclodextrin based nanosponge tablets
467 with a controlled release system were developed. The compositions of formulations were
468 Febux: nanosponge = 1:1 (composition of nanosponges were β -cyclodextrin: Diphenyl
469 carbonate = 1:2, 1:4, 1:6, 1:8, and 1:10). The dissolution (%) and *in vitro* release pattern was
470 lower than that of SD3 and Goutifade tablet[®] (commercial products). Still, oral bioavailability
471 was twice as high as that of Goutifade tablet[®] in Sprague Dawley rats (Amin et al., 2020). In
472 the third previous study, the dissolutions (%) of Febux solid dispersions (Febux:PVP
473 K30[®]:P188[®] = 1:3:3 and Febux:PVP K30[®]:P407[®] = 1:3:3) were below 50% and 95% in DW
474 and pH 6.8 buffer, respectively. Compared to the SD3, it showed significantly lower dissolution
475 results ($100.4 \pm 0.3\%$ and $98.8 \pm 1.2\%$ in DW and pH 6.8 buffer, respectively.) The AUC_{0-24h}
476 ($\mu\text{h/mL}$) Febux solid dispersions showed 1.5-fold (P188[®] formulation) and 1.4-fold (P407[®]
477 formulation) higher than pure Febux in Sprague–Dawley (SD) rats (Tang et al., 2018).

478 The our SD3 formulation is expected to have high oral bioavailability compared to the
479 above papers by considering the high dissolution (%) and *in vitro* release pattern. Basically,

480 when drug solubilization increases the total weight in the formulation. However, the SD3
481 formulation is considered to have no problem with dosage compliance even when compared to
482 the existing 260mg reference drug with a total weight of 200mg.

483 **In Fig. 6**, the melting peaks of SD3 and PM3 were observed in the order of P407[®],
484 meglumine, and mannitol, whereas the melting peak of pure Febux was not observed in DSC
485 data. Therefore, it was difficult to distinguish the thermal changes in SD3. A difference
486 between PM3 and SD3 was that C=O stretching band shifted from 1676.1 cm⁻¹ (PM3) to
487 1628.3 cm⁻¹ (SD3). This result indicated that Febux and the excipient have hydrogen bonds.
488 Thus, the dissolution (%) of Febux in SD3 may have improved because of the chemical
489 interactions between Febux and meglumine. The XRD patterns of PM3 and SD3 were similar
490 and mostly coincided with meglumine peaks. The peak intensity of Febux was considered weak
491 and could not be observed for PM3 and SD3.

492 The composition of Febux-solid dispersion (SD) formulations was excluded one by one to
493 identify the main factors in **Fig. 7**. To confirm this, it was confirmed while excluding the
494 excipient of the SD3 formulation by one. The SD3 (Febux, meglumine, P407[®], and mannitol
495 in a 1:2:1:1 ratio), SD3-1 (Febux, meglumine, P407[®], and mannitol in a 1:0:1:1 ratio), SD3-2
496 (Febux, meglumine, P407[®], and mannitol in a 1:2:0:1 ratio), and SD3-3 (Febux, meglumine,
497 P407[®], and mannitol in a 1:2:1:0 ratio) were prepared in the same manner as described above
498 (**Table 3**). In most formulations, stability was maintained but tended to decrease by
499 approximately 10% in SD3-3 for 6 months. The results suggest that mannitol is effective in
500 maintaining stability. Moreover, meglumine had the highest solubilizing effect on Febux, and
501 the effect of P407[®] or mannitol was insignificant. In previous studies, tadalafil-SD formulation
502 was performed to find the main factors. PVP/VA S-630 was important to improve the pre-
503 dissolution (%) of tadalafil in tadalafil-SD formulation (consist of chitosan, Aerosil[®]200 and
504 PVP/VA S-630) (Sohn and Choi, 2021b). Additionally, the naftopidil-SD formulation was

505 confirmed as the key factor. Fumaric acid was essential to increase the pre-dissolution (%) of
506 naftopidil in naftopidil-SD formulation (consisting of fumaric acid, chitosan, and US2[®]) (Sohn
507 et al., 2021a). Also, the crystallinity of Febux in SD3 was stable for 6 months, as shown in **Fig.**
508 **6c**. In previous studies, the Febux-SD formulations did not undergo stability tests (Amin et al.,
509 2020; El Shenawy et al., 2019; Jagia et al., 2022; Kaur et al., 2020; Moinuddin et al., 2020;
510 Patel et al., 2021).

511 **5. Conclusion**

512 In this study, the solubilization of Febux, a BCS II drug, was successfully developed with
513 meglumine with the SD technique. The SD3 formulation consisted of Febux, meglumine,
514 P407[®], and mannitol in a 1:2:1:1 weight ratio. The dissolution (%) of Febux in SD3 formulation
515 significantly enhanced the dissolution (%) of Febux compared to the Feburic[®] tab in all media.
516 Moreover, in *in vitro* release, the SD3 formulation showed a faster release pattern than the
517 Feburic[®] tab in the pH shifting method. In the *in vitro* permeability test, the SD3 formulation
518 showed a higher value than to Feburic[®] tab in four membrane types.

519 Furthermore, in SD3 formulation, hydrogen bonding between Febux and meglumine was
520 confirmed. It is thought that the dissolution (%) and *in vitro* release of Febux in SD3 was
521 improved owing to the change in this chemical interaction. The order of meglumine > P407[®]
522 > mannitol in the SD3 formulation was important for improving the pre-dissolution (%) of
523 Febux. The SD3 formulation was maintained for 6 months. This research developed a
524 formulation by applying meglumine to Febux for the first time. The SD3 formulation improved
525 dissolution (%), drug release (%), and stability compared to commercial products (Feburic[®]
526 tab). Moreover, it is a formulation that does not increase total weight compared to the
527 commercial products (Feburic[®] tab). Therefore, it seems that there is no problem with patient

528 compliance. Based on these results, the SD3 formulation is able to increase the oral
529 bioavailability of Febux in animal or human studies.

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539 **Declaration of competing interest**

540 The authors declare that they have no known competing financial interests or personal
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542

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620

Table 1

621

Composition of the febuxostat (Febux) F formulations (mg, per batch).

	F1	F2	F3	F4	F5	F6	F7	F8	F9	F10	F11	F12
Febux	40	40	40	40	40	40	40	40	40	40	40	40
Meglumine	80	80	80	80	80	80	80	80	80	80	80	80
P188 [®]	40	-	-	-	-	-	-	-	-	-	-	-
P407 [®]	-	40	-	-	-	-	-	-	-	-	-	-
PEG6000	-	-	40	-	-	-	-	-	-	-	-	-
Soluplus [®]	-	-	-	40	-	-	-	-	-	-	-	-
IR [®]	-	-	-	-	40	-	-	-	-	-	-	-
PVP/VA S630	-	-	-	-	-	40	-	-	-	-	-	-
K12 [®]	-	-	-	-	-	-	40	-	-	-	-	-
K17 [®]	-	-	-	-	-	-	-	40	-	-	-	-
K30 [®]	-	-	-	-	-	-	-	-	40	-	-	-
K90 [®]	-	-	-	-	-	-	-	-	-	40	-	-
TPGS	-	-	-	-	-	-	-	-	-	-	40	-
HS 15 [®]	-	-	-	-	-	-	-	-	-	-	-	40
Total	160	160	160	160	160	160	160	160	160	160	160	160

622

623

624 **Table 2**
 625 Composition of the febuxostat (Febux) SD formulations (mg, per batch).

	SD1	SD2	SD3	SD4	SD5	SD6	SD7	SD8	SD9
Febux	40	40	40	40	40	40	40	40	40
Meglumine	80	80	80	80	80	80	80	80	80
P407®	40	40	40	40	40	40	40	40	40
Aerosil®200	40	-	-	-	-	-	-	-	-
Aerosil®300	-	40	-	-	-	-	-	-	-
Mannitol	-	-	40	-	-	-	-	-	-
Flowlac	-	-	-	40	-	-	-	-	-
MCC	-	-	-	-	40	-	-	-	-
DCP-A	-	-	-	-	-	40	-	-	-
DCP-D	-	-	-	-	-	-	40	-	-
US2	-	-	-	-	-	-	-	40	-
UFL2	-	-	-	-	-	-	-	-	40
Total	200	200	200	200	200	200	200	200	200

626

628 **Table 3**

629 Composition of the febuxostat (Febux) SD formulations (mg, per batch).

	SD3	SD3-1	SD3-2	SD3-3
Febux	40	40	40	40
Meglumine	80	-	80	80
P407 [®]	40	40	-	40
Mannitol	40	40	40	-
Total	200	120	160	160

631 **Figure Captions**

632 **Fig. 1. Pre-dissolution study of the B formulations.** Pre-dissolution (%) of B formulations
633 (B1–24) in distilled water was tested at 37 ± 1 °C for 60 min. Data are expressed as mean
634 \pm standard deviation (sd, n = 3).

635 **Fig. 2. Pre-dissolution study of the F and SD (solid dispersion) formulations.** Pre-
636 dissolution (%) of F formulations (F1-12) in distilled water was tested at 37 ± 1 °C for 60
637 min (a). Pre-dissolution (%) of SD formulations (SD1-9) in distilled water was tested at
638 37 ± 1 °C for 60 min (b).

639 **Fig. 3. Dissolution study of the optimal formulation.** Dissolution (%) of pure Febux,
640 Feburic[®] tab, physical mixture (PM3), and solid dispersion (SD3) was tested in pH 1.2
641 (a), distilled water (DW) (b), and pH 6.8 buffer (c) at 37 ± 1 °C for 60 min. Data are
642 expressed as mean \pm standard deviation (sd, n = 6).

643 **Fig. 4. *In vitro* drug release test.** The *in vitro* drug release test of pure Febux, Feburic[®] tab,
644 physical mixture (PM3), and solid dispersion (SD3) was evaluated in pH shifting method.
645 Data are expressed as mean \pm standard deviation (sd, n = 3).

646 **Fig. 5. *In vitro* permeability test.** The *in vitro* permeability test of Feburic[®] tab, physical
647 mixture (PM3), and solid dispersion (SD3) was evaluated with PBS (pH 7.4) in Franz
648 diffusion cell system. Data are expressed as mean \pm standard deviation (sd, n = 3).

649 **Fig. 6. Physicochemical properties.** DSC images of pure Febux, Feburic[®] tab, physical
650 mixture (PM3), and solid dispersion (SD3) (a). FT-IR spectra of pure Febux, Feburic[®] tab,
651 physical mixture (PM3), and solid dispersion (SD3) (b). PXRD images of pure Febux,
652 Feburic[®] tab, physical mixture (PM3), and solid dispersion (SD3) (c).

653 **Fig. 7. Stability test.** Composition of Febux-solid dispersion (SD) formulations by excluding
654 each variable to identify main factors. SD3 (Febux, meglumine, P407[®] and mannitol in a
655 1:2:1:1), SD3-1 (Febux, meglumine, P407[®] and mannitol in a 1:0:1:1), SD3-2 (Febux,
656 meglumine, P407[®] and mannitol in a 1:2:0:1) and SD3-3 (Febux, meglumine, P407[®] and
657 mannitol in a 1:2:1:0)(a). Moreover, stability test was performed for 6 months. The pre-
658 dissolution (%) of Febux in SD3 formulations was performed in DW at 37 ± 1 °C at 60 min
659 (b). Graph represents the mean \pm standard deviation (n = 3).

660

Figure

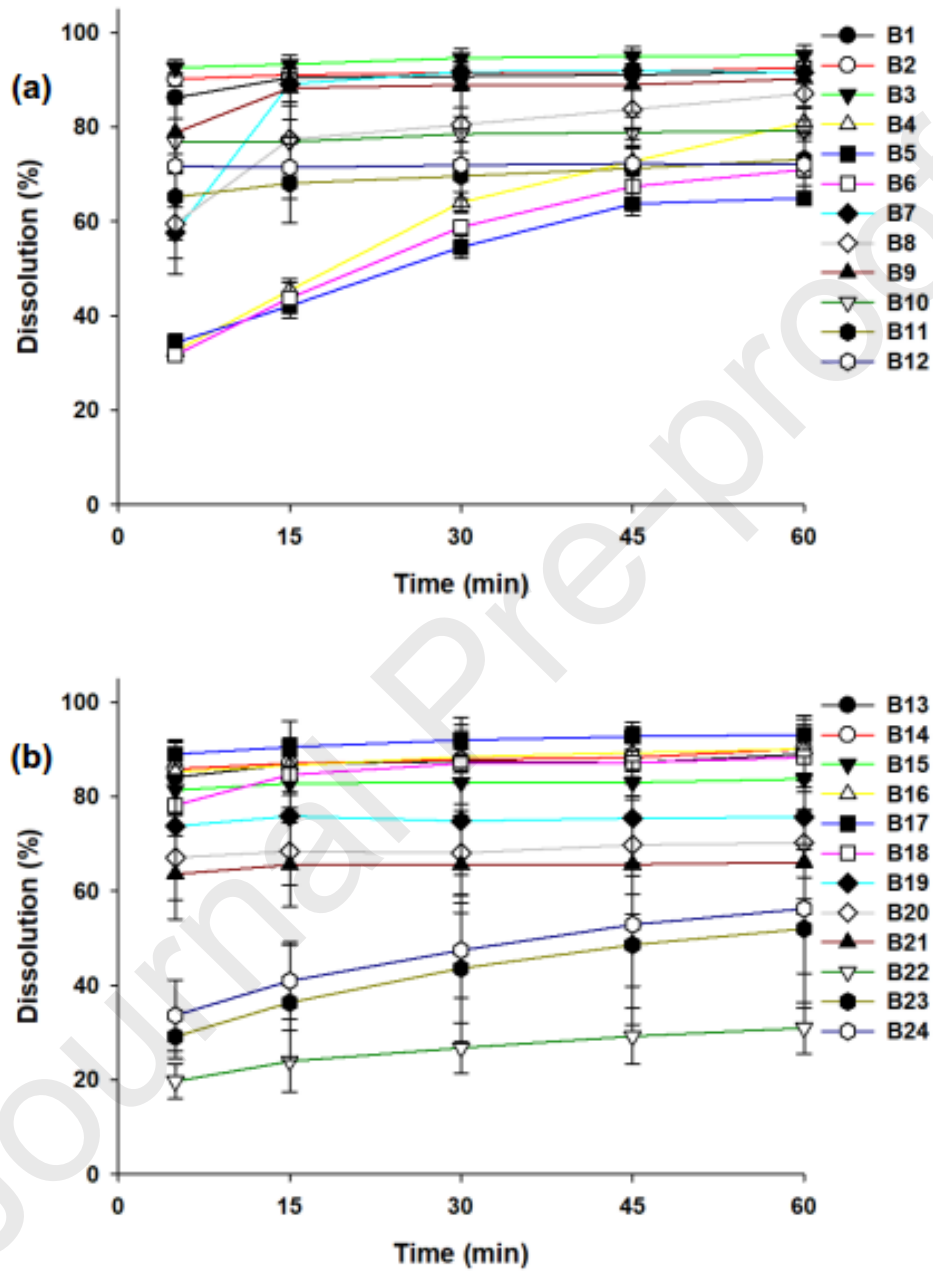
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Figure. 1

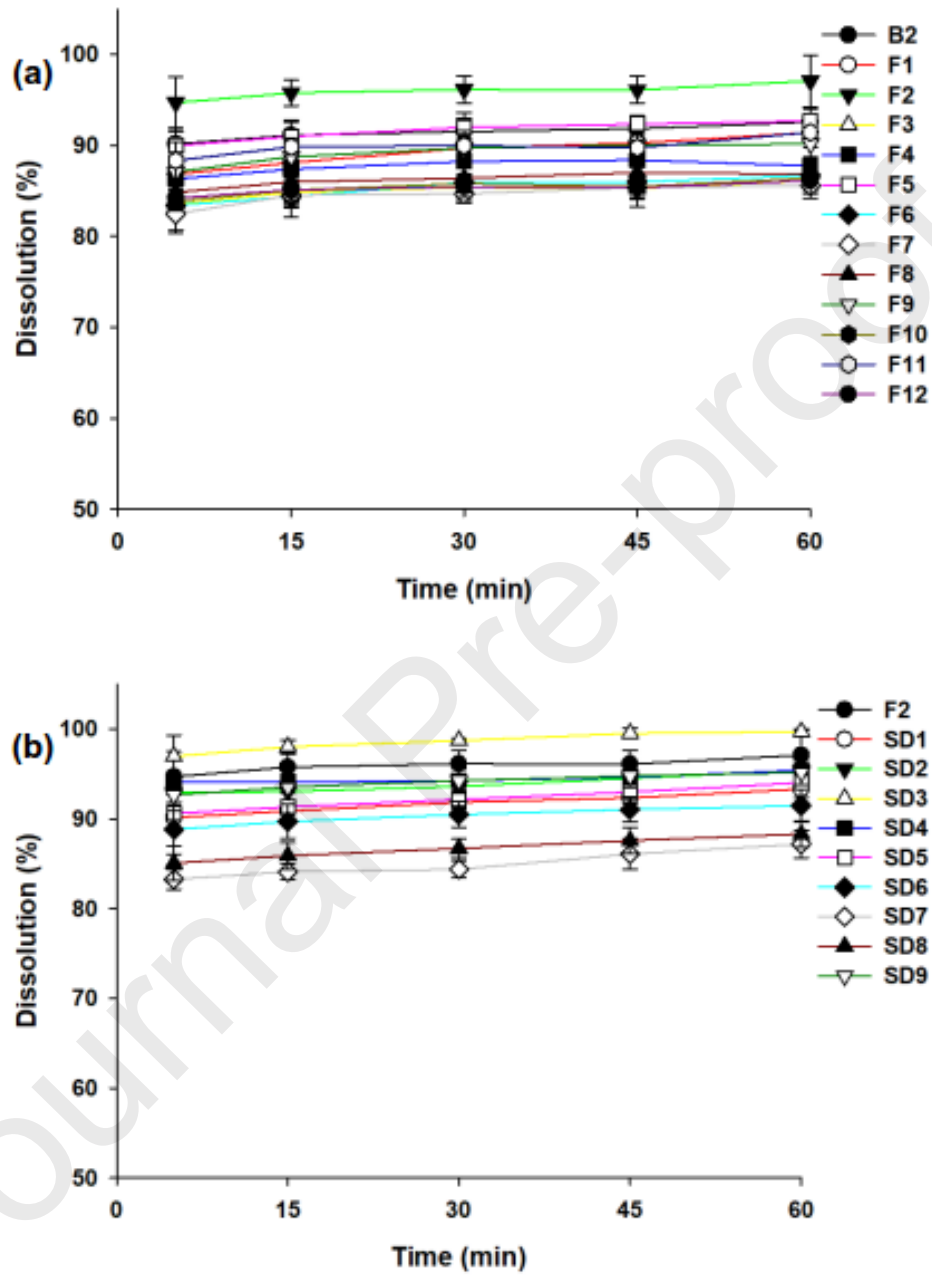


Figure. 2

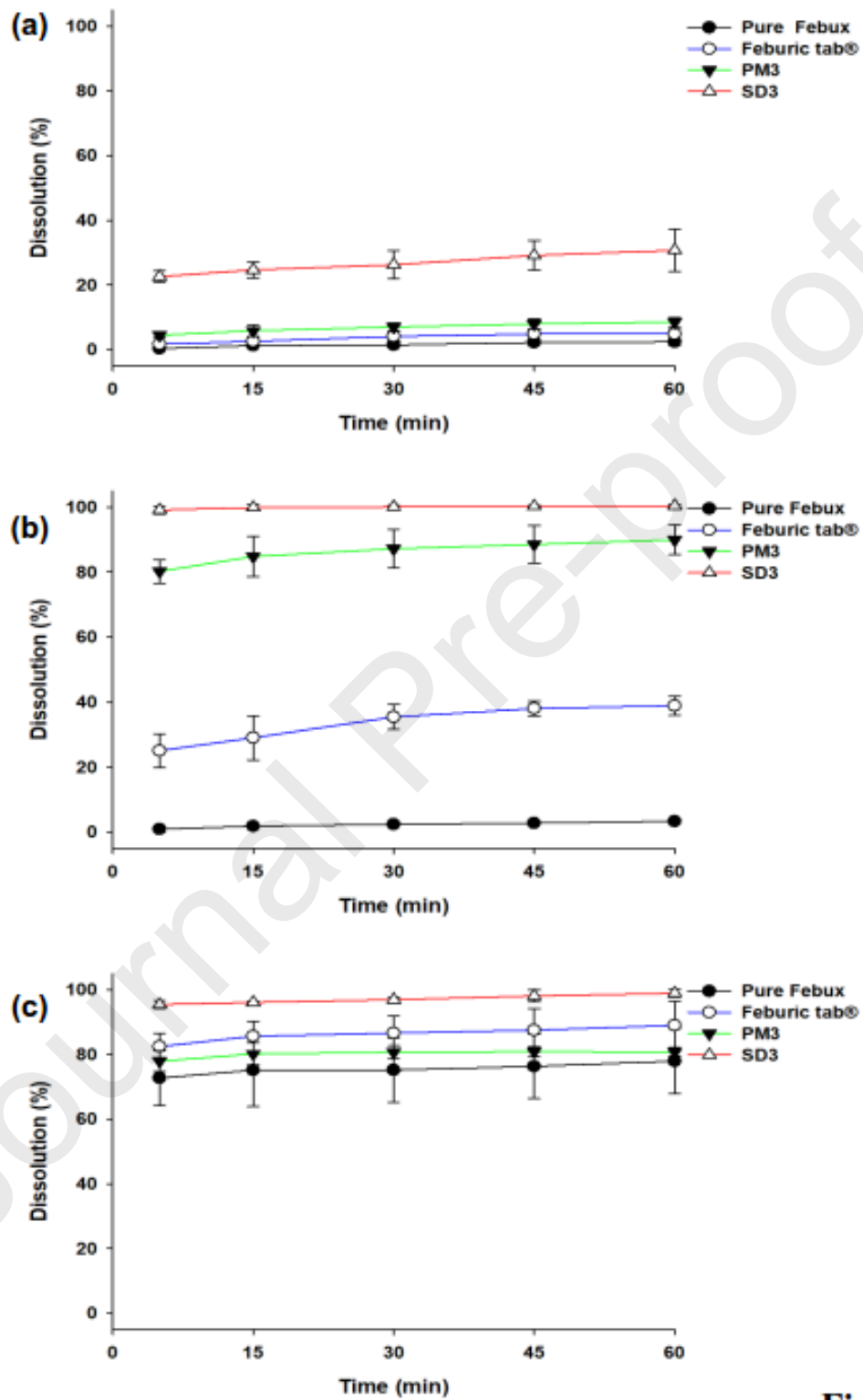


Figure. 3

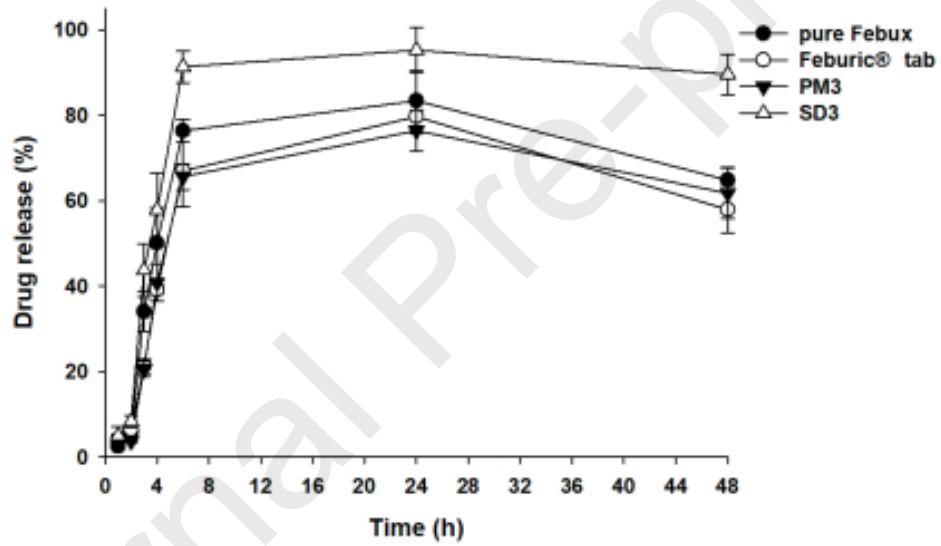


Figure. 4

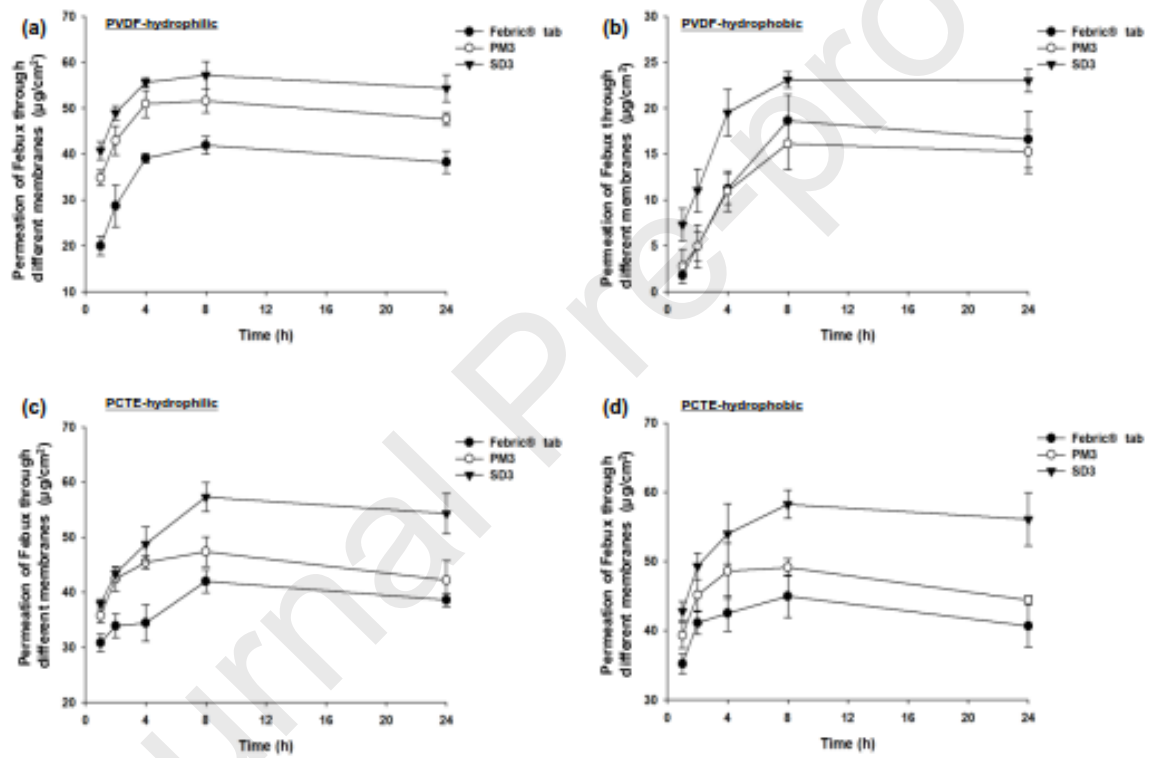


Figure. 5

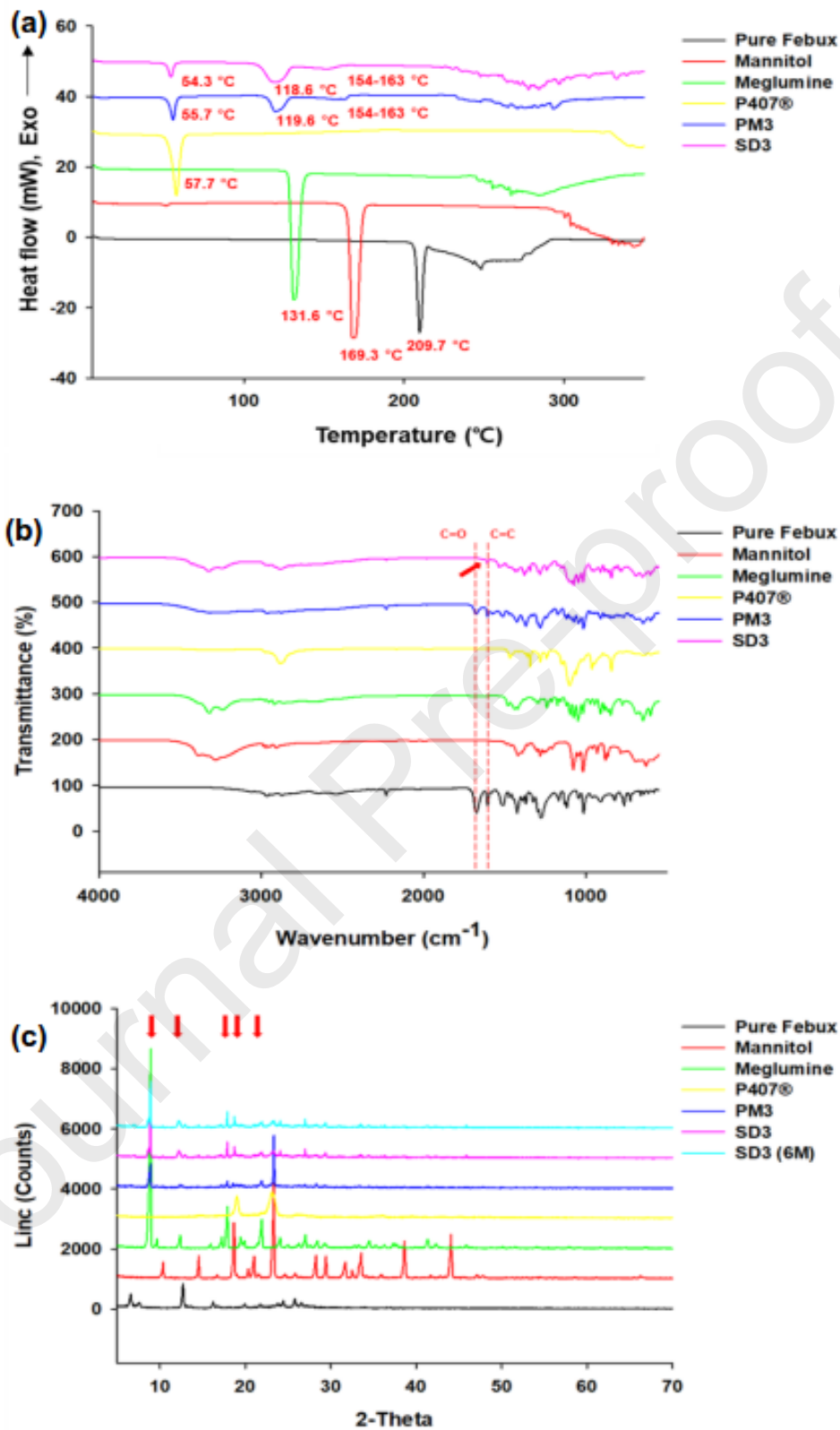


Figure. 6

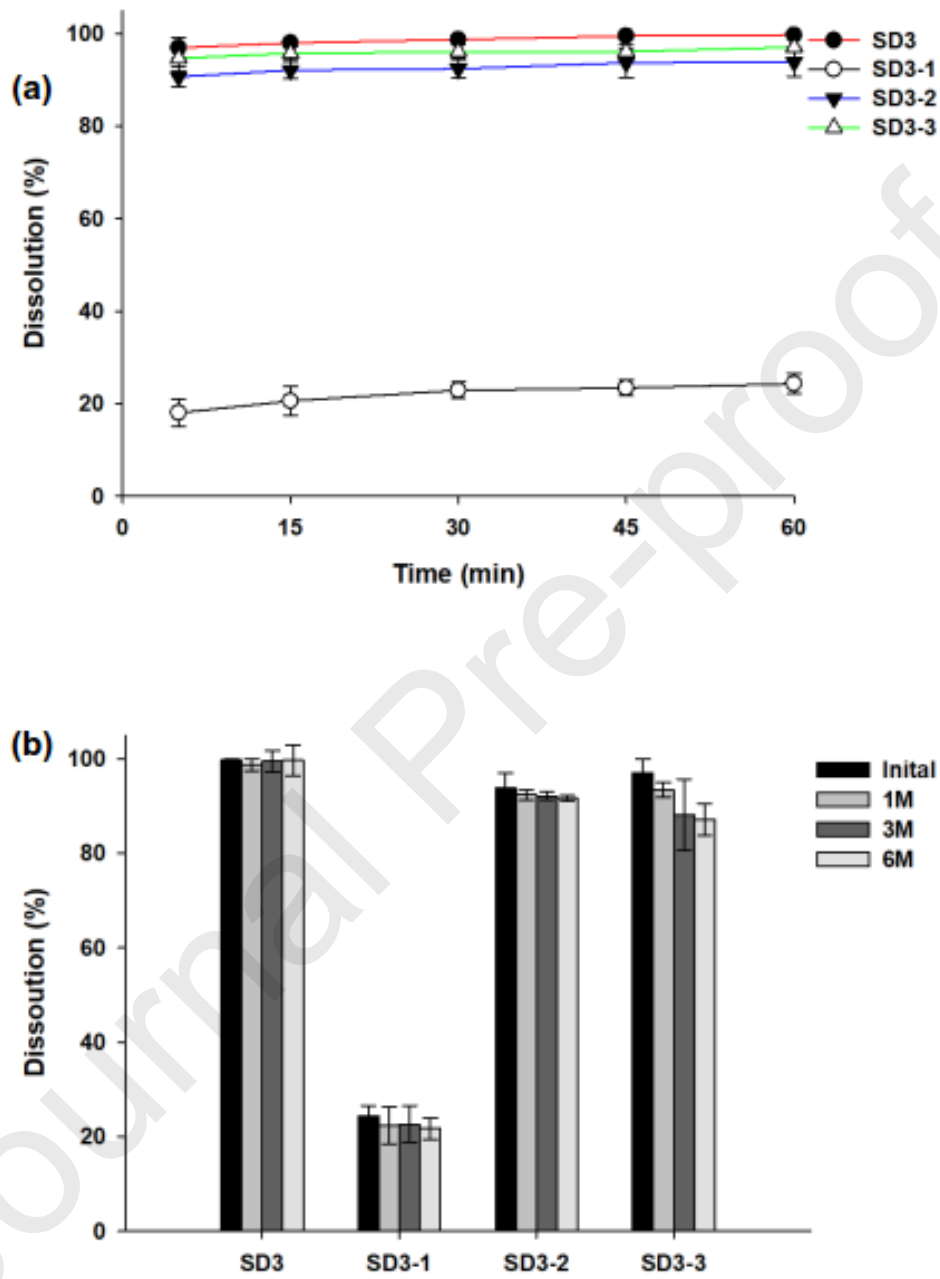


Figure. 7